

**Registry No.**—Propionic acid, 79-09-4; hydrochloric acid, 7647-01-0.

**Acknowledgment.**—This work was supported by NSF Grant BG-7033X and NIH Grant HD-01262 to Professor M. D. Kamen, NIH Grant GM-10928 to Professor J. Kraut, and NIH Grant AM 14879-02.

### The Origin of the $[M - 56] \cdot^+$ Ion in the Mass Spectra of Trimethylsilyl Ethers of Dehydroepiandrosterone and Related Compounds

C. J. W. BROOKS

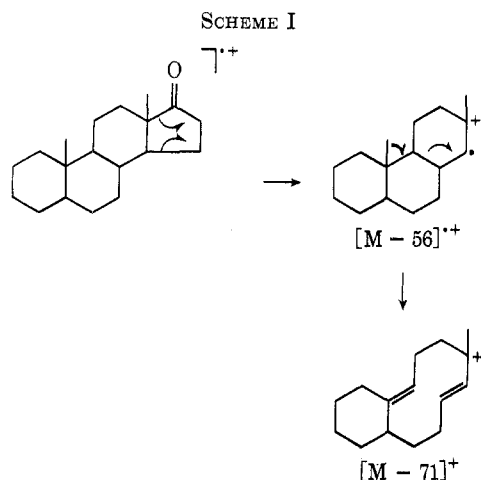
Department of Chemistry, The University, Glasgow, G12 8QQ, Scotland

D. J. HARVEY AND B. S. MIDDLEDITCH\*

Institute for Lipid Research, Baylor College of Medicine, Houston, Texas 77025

Received March 3, 1972

The mass spectra of many 16- and 17-keto steroids contain ions  $[M - 56] \cdot^+$ , the formation of which has been ascribed to cleavages of the bonds C-13/17 and C-14/15.<sup>1</sup> These ions are often accompanied by ions  $[M - 71]^+$  formed by subsequent loss of a methyl radical<sup>2</sup> (Scheme I). During a survey of the mass



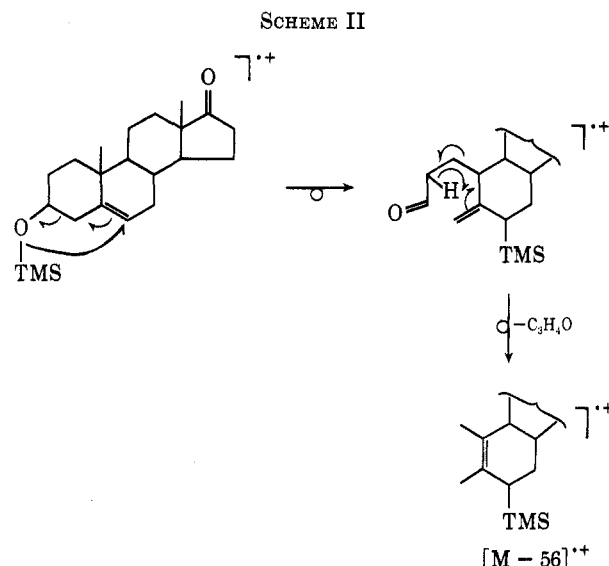
spectra of trimethylsilyl (TMS) ethers of a number of  $\Delta^5$ -3 $\beta$ -hydroxy steroids it was found that  $[M - 56] \cdot^+$  ions, unaccompanied by  $[M - 71]^+$  ions, were present in the spectra of 16 and 17 ketones. It has been demonstrated that these  $[M - 56] \cdot^+$  ions are formed by electron-impact-induced rearrangement, and not by D-ring cleavage.

(1) G. von Unruh and G. Spiteller, *Tetrahedron*, **26**, 3289 (1970), and references cited therein.

(2) L. Tökés, R. T. LaLonde, and C. Djerassi, *J. Org. Chem.*, **32**, 1012 (1967).

The first indication of a duality of mechanisms for the formation of such ions was the presence of  $[M - 56] \cdot^+$  ions in the spectra of the TMS ethers of 15,15-, 17,17-*d*<sub>4</sub>-3 $\beta$ -hydroxyandrost-5-en-16-one, 16,16-*d*<sub>2</sub>-3 $\beta$ -hydroxyandrost-5-en-17-one, and 9,12,12,16,16-*d*<sub>5</sub>-3 $\beta$ -hydroxyandrost-5-ene-11,17-dione.<sup>3</sup> When the 17-oxo group of the TMS ether of 3 $\beta$ -hydroxyandrost-5-en-17-one (dehydroepiandrosterone, DHEA) was selectively replaced<sup>4</sup> by <sup>18</sup>O, this atom was found to be retained in the  $[M - 56] \cdot^+$  ion. All nine deuterium atoms of the *d*<sub>9</sub>-TMS ether<sup>5</sup> of DHEA were also retained in the  $[M - 56] \cdot^+$  ion.

High resolution mass measurement, carried out on the spectrum of the TMS ether of DHEA, showed that the particle eliminated had the composition C<sub>3</sub>H<sub>4</sub>O (found for ion of nominal *m/e* 304, 304.2198; calcd for C<sub>3</sub>H<sub>32</sub>OSi, 304.2222). The oxygen atom must, therefore, originate from the 3 position, and it seems likely that the  $[M - 56] \cdot^+$  ions of these steroids are formed by a mechanism similar to that proposed for the formation of the  $[M - 129]^+$  ion, but with initial transfer of the TMS group. This may proceed *via* a double (silyl and conventional) McLafferty-type rearrangement, as in Scheme II. Because of the relatively large separation



of C-3 and C-6, the silyl rearrangement is presumed to take place in a stepwise manner.

It should be noted that the TMS ethers of the saturated steroid 3 $\beta$ -hydroxy-5 $\alpha$ -androst-17-one and its 16,16-*d*<sub>2</sub> analog give rise, respectively, to ions  $[M - 56] \cdot^+$  and  $[M - 58] \cdot^+$ , indicating that such ions are formed by D-ring cleavage as illustrated in Scheme I.<sup>6</sup>

(3) These deuterated analogs were prepared by *in transitu* labeling during GC-MS: G. M. Anthony and C. J. W. Brooks, *Chem. Commun.*, 200 (1970).

(4) A. M. Lawson, F. A. J. M. Leemans, and J. A. McCloskey, *Steroids*, **14**, 603 (1969).

(5) J. A. McCloskey, R. N. Stillwell, and A. M. Lawson, *Anal. Chem.*, **40**, 233 (1968).

(6) This work was supported by grants from the Medical Research Council (to C. J. W. B.), Science Research Council (B/SR/2398 to C. J. W. B. and G. Eglington), NIH (GM-13901), and the Robert A. Welch Foundation (Q-125).